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the atomic number z is the number of protons in the nucleus the mass number a is the total number of protons and neutrons in the atom number of neutrons a z isotopes isotopes are atoms with the same number of protons but different numbers of neutrons isotopes have similar chemical properties because they have the same electronic structure, **5 atomic structure and the periodic table bergen** - atomic mass number is closer to the average atomic mass of chlorine a 35 amu b 37 amu chapter 5 atomic structure and the periodic table continued an atom of neon 22 has two more neutrons in its nucleus than an atom of neon 20 isotopes true b c a carbon 12 12 amu false the values of atomic masses measured in grams are inconveniently small and, **sam teachers guide atomic structure concord consortium** - atomic structure page 4 atomic model building o clarify what defines an ion and use examples to review o link to other sam activities chemical bonding students need to understand the structure of the atom to understand chemical bonding and reactions in the atomic world, **4 atomic structure guided answers pearson education 238377** - 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